Pulizia Proteine - Protein Cleaning Solution

Revision nr.2 Dated 25/05/2023 Printed on 25/05/2023 Page n. 1 / 14

Replaced revision:1 (Dated 19/12/2022)

Safety Data Sheet

According to Annex II to REACH - Regulation 2020/878 and to Annex II to UK REACH

SECTION 1. Identification of the substance/mixture and of the company/undertaking

1.1. Product identifier

Code: Pulizia Proteine

Product name Chemical name and synonym Protein Cleaning Solution Pepsin - hydrochloric acid

UFI: **1500-X0XT-A00D-Y3HG**

1.2. Relevant identified uses of the substance or mixture and uses advised against

Intended use Laboratory reagent. Applies to codes 32208063, 32208093, 32308123

1.3. Details of the supplier of the safety data sheet

Name GIORGIO BORMAC srl Full address via della meccanica, 25

District and Country 41012 Carpi (MO)

Italia

Tel. +39 059 653274 Fax +39 059 653282

e-mail address of the competent person

responsible for the Safety Data Sheet sds@giorgiobormac.com

Supplier: GIORGIO BORMAC srl

1.4. Emergency telephone number

For urgent inquiries refer to +44 121 507 4123

SECTION 2. Hazards identification

2.1. Classification of the substance or mixture

The product is classified as hazardous pursuant to the provisions set forth in (EC) Regulation 1272/2008 (CLP) (and subsequent amendments and supplements). The product thus requires a safety datasheet that complies with the provisions of (EU) Regulation 2020/878

Any additional information concerning the risks for health and/or the environment are given in sections 11 and 12 of this sheet.

Hazard classification and indication:

Substance or mixture corrosive to metals, category 1 H290 May be corrosive to metals.

Skin corrosion, category 1B H314 Causes severe skin burns and eye damage.

Serious eye damage, category 1 H318 Causes serious eye damage.

2.2. Label elements

Hazard labelling pursuant to EC Regulation 1272/2008 (CLP) and subsequent amendments and supplements.

Hazard pictograms:



Signal words: Danger

Hazard statements:

H290 May be corrosive to metals.

H314 Causes severe skin burns and eye damage.

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SECTION 2. Hazards identification .../>>

Precautionary statements:

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to

do. Continue rinsing.

P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].

P260 Do not breathe dust / fume / gas / mist / vapours / spray.

P280 Wear protective gloves/ protective clothing / eye protection / face protection.

P310 Immediately call a POISON CENTER / doctor / . . .

P264 Wash . . . thoroughly after handling.

Contains: HYDROCHLORIC ACID

Mercury chloride ico

2.3. Other hazards

On the basis of available data, the product does not contain any PBT or vPvB in percentage ≥ than 0,1%.

The product does not contain substances with endocrine disrupting properties in concentration ≥ 0.1%.

SECTION 3. Composition/information on ingredients

3.2. Mixtures

Contains:

FC

Identification x = Conc. % Classification (EC) 1272/2008 (CLP)

HYDROCHLORIC ACID

INDEX 017-002-01-X $0.2 \le x < 0.4$ Met. Corr. 1 H290, Skin Corr. 1B H314, Eye Dam. 1 H318, STOT SE 3 H335,

Classification note according to Annex VI to the CLP Regulation: B

Met. Corr. 1 H290: ≥ 0,1%, Skin Corr. 1B H314: ≥ 25%, Skin Irrit. 2 H315: ≥ 10%, Eye Dam. 1 H318: ≥ 25%, Eye Irrit. 2 H319: ≥ 10%, STOT SE 3 H335: ≥

10%

CAS 7647-01-0

REACH Reg. 01-2119484862-27-xxxx

231-595-7

Mercury chloride ico

INDEX 080-010-00-X 0 ≤ x < 0,01 Muta. 2 H341, Repr. 2 H361, Acute Tox. 2 H300, STOT RE 1 H372, Skin Corr.

1B H314, Eye Dam. 1 H318, Aquatic Acute 1 H400 M=100, Aquatic Chronic 1

H410 M=1

EC 231-299-8 LD50 Oral: 37 mg/kg

CAS 7487-94-7

The full wording of hazard (H) phrases is given in section 16 of the sheet.

SECTION 4. First aid measures

4.1. Description of first aid measures

In the event of an accident or to malaise, consult the doctor (if possible, show him the label). In case of frantic breath, administer oxygen. Make sure that the medical staff is aware of the materials involved, and take the necessary precautions to protect themselves.

Eyes: wash immediately and abundantly with water for at least 15 minutes and call the doctor immediately.

Leather: Remove the clothes immediately and wash the skin with a lot of water. Dab with polyetilenglicle 400. Consult a doctor.

Ingestion: make a lot of water drink. If possible, avoid vomiting (dangerous of laceration). In case of vomiting, keep your head down to prevent this from penetrating the lungs. Consult a doctor.

Inhalation: bring the subject to the open air and keep it at rest. If breathing ceases or is difficult, practice artificial respiration by adopting adequate precautions for the rescuer. Consult a doctor.

4.2. Most important symptoms and effects, both acute and delayed

HYDROCHLORIC ACID

Acute effects-dependent effects. Cute: irritation, burns, ulcer Eyes: irritation, corneal damage

Nose: irritation First airways: irritation Lungs: irritation

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SECTION 4. First aid measures .../>>

Digerent apparatus: if ingested retrostern and epigastric pain, hematemesis

Chronic effects.

Cute: irritation, depigmentation, dry skin, epilation

Eyes: irritation Nose: irritation First airways: irritation Lungs: irritation

Mercury chloride ico

Contact with the skin: toxic for skin absorption, can give rise to burns and burns

Contact with eyes: it causes burns and irritation

Ingestion: it can be fatal if ingested, it causes burns and irritation

Inhalation: it can be harmful if inhaled, the material is highly destructive for the mucous membranes and for the upper respiratory tract

4.3. Indication of any immediate medical attention and special treatment needed

Consult a doctor in case of contact with the substance

SECTION 5. Firefighting measures

The product is not flammable and does not feed the flames.

5.1. Extinguishing media

SUITABLE EXTINGUISHING EQUIPMENT

The extinguishing equipment should be of the conventional kind: carbon dioxide, foam, powder and water spray.

UNSUITABLE EXTINGUISHING EQUIPMENT

None in particular.

5.2. Special hazards arising from the substance or mixture

HAZARDS CAUSED BY EXPOSURE IN THE EVENT OF FIRE

Do not breathe combustion products.

Mercury chloride ico

In combustion he emits toxic fumes of hydrogen chloride and mercury oxides.

5.3. Advice for firefighters

GENERAL INFORMATION

Use jets of water to cool the containers to prevent product decomposition and the development of substances potentially hazardous for health. Always wear full fire prevention gear. Collect extinguishing water to prevent it from draining into the sewer system. Dispose of contaminated water used for extinction and the remains of the fire according to applicable regulations.

SPECIAL PROTECTIVE EQUIPMENT FOR FIRE-FIGHTERS

Normal fire fighting clothing i.e. fire kit (BS EN 469), gloves (BS EN 659) and boots (HO specification A29 and A30) in combination with self-contained open circuit positive pressure compressed air breathing apparatus (BS EN 137).

SECTION 6. Accidental release measures

6.1. Personal precautions, protective equipment and emergency procedures

Block the leakage if there is no hazard.

Wear suitable protective equipment (including personal protective equipment referred to under Section 8 of the safety data sheet) to prevent any contamination of skin, eyes and personal clothing. These indications apply for both processing staff and those involved in emergency procedures.

6.2. Environmental precautions

The product must not penetrate into the sewer system or come into contact with surface water or ground water.

6.3. Methods and material for containment and cleaning up

Collect the leaked product into a suitable container. Evaluate the compatibility of the container to be used, by checking section 10. Absorb the remainder with inert absorbent material.

Make sure the leakage site is well aired. Contaminated material should be disposed of in compliance with the provisions set forth in point 13.

6.4. Reference to other sections

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Any information on personal protection and disposal is given in sections 8 and 13.

SECTION 7. Handling and storage

7.1. Precautions for safe handling

During manipulation, use the means of protection indicated in point 8 of this card and the procedures below: do not smoke, do not eat, not drink during manipulation; Use particular caution in manipulation, to avoid any exposure to the product. Predict accurate ventilation/suction in the workplace; Wash your hands thoroughly after manipulation and at the end of the round; The shower is recommended if manipulated in large quantities.

7.2. Conditions for safe storage, including any incompatibilities

In the storage of the preparation use the precautions below:

- keep in mind the chemical-physical characteristics of the preparation to avoid possible interactions with other products
- Keep the containers hermetically closed and in a cool and ventilated place, not exposed to direct sunlight. Keep away from strong bases.

7.3. Specific end use(s)

Information not available

SECTION 8. Exposure controls/personal protection

8.1. Control parameters

Regulatory References:

ITA Italia Decreto Legislativo 9 Aprile 2008, n.81

EU OEL EU Directive (EU) 2022/431; Directive (EU) 2019/1831; Directive (EU) 2019/130; Directive (EU)

2019/983; Directive (EU) 2017/2398; Directive (EU) 2017/164; Directive 2009/161/EU; Directive

2006/15/EC; Directive 2004/37/EC; Directive 2000/39/EC; Directive 98/24/EC; Directive

91/322/EEC.

TLV-ACGIH ACGIH 2021

				HYDROC	HLORIC ACIE)			
Threshold Limit	Value								
Type	Country	TWA/8h		STEL/15min		Remarks / Observations			
		mg/m3	ppm	mg/m3	ppm				
VLEP	ITA	8	5	15	10				
OEL	EU	8	5	15	10				
TLV-ACGIH		2,98 (C)		A4 URT IRR					
lealth - Derived	no-effect lev	el - DNEL /	DMEL						
Effects on co		cts on consi	consumers			Effects on workers			
Route of expos	sure Acu	te Acı	ute	Chronic	Chronic	Acute local	Acute	Chronic	Chronic
	loca	l sys	temic	local	systemic		systemic	local	systemic
Inhalation	15			8		15		8	
	mg/	m3		mg/m3		mg/m3		mg/m3	

				Mercury	chloride icc	
Threshold Lin	nit Value			moroury	omorrae rec	•
Type	Country	TWA/8h		STEL/15	min	Remarks / Observations
.,,,,,		mg/m3	ppm	mg/m3	ppm	Trainer, executations
OEL	EU	0,02	FF		FF	Come Hg

Legend

(C) = CEILING; INHAL = Inhalable Fraction; RESP = Respirable Fraction; THORA = Thoracic Fraction.

VND = hazard identified but no DNEL/PNEC available ; NEA = no exposure expected ; NPI = no hazard identified ; LOW = low hazard

; MED = medium hazard ; HIGH = high hazard.

HYDROCHLORIC ACID

Sampling methods

The following sampling methods for the substances mentioned in the previous tables are suggested.

 $http://amcaw.ifa.dguv.de/substance/methoden/093-\ Hydrogen_chloride_2016.pdf.$

Mercury chloride ico

Sampling methods

The following sampling methods for the substances mentioned in the previous tables are suggested.

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http://amcaw.iifa.dguv.de/substance/methoden/075-l-mercury.pdf

8.2. Exposure controls

As the use of adequate technical equipment must always take priority over personal protective equipment, make sure that the workplace is well aired through effective local aspiration.

When choosing personal protective equipment, ask your chemical substance supplier for advice.

Personal protective equipment must be CE marked, showing that it complies with applicable standards.

Provide an emergency shower with face and eye wash station.

HAND PROTECTION

Protect hands with category III work gloves (see standard EN 374).

The following should be considered when choosing work glove material: compatibility, degradation, failure time and permeability.

The work gloves' resistance to chemical agents should be checked before use, as it can be unpredictable. The gloves' wear time depends on the duration and type of use.

SKIN PROTECTION

Wear category II professional long-sleeved overalls and safety footwear (see Regulation 2016/425 and standard EN ISO 20344). Wash body with soap and water after removing protective clothing.

EYE PROTECTION

Wear airtight protective goggles (see standard EN 166).

RESPIRATORY PROTECTION

If the threshold value (e.g. TLV-TWA) is exceeded for the substance or one of the substances present in the product, use a mask with a type B filter whose class (1, 2 or 3) must be chosen according to the limit of use concentration. (see standard EN 14387). In the presence of gases or vapours of various kinds and/or gases or vapours containing particulate (aerosol sprays, fumes, mists, etc.) combined filters are required. Respiratory protection devices must be used if the technical measures adopted are not suitable for restricting the worker's exposure to the threshold values considered. The protection provided by masks is in any case limited.

If the substance considered is odourless or its olfactory threshold is higher than the corresponding TLV-TWA and in the case of an emergency, wear open-circuit compressed air breathing apparatus (in compliance with standard EN 137) or external air-intake breathing apparatus (in compliance with standard EN 138). For a correct choice of respiratory protection device, see standard EN 529. ENVIRONMENTAL EXPOSURE CONTROLS

The emissions generated by manufacturing processes, including those generated by ventilation equipment, should be checked to ensure compliance with environmental standards.

Information

SECTION 9. Physical and chemical properties

9.1. Information on basic physical and chemical properties

Properties Value Appearance clear liquid Colour colourless Odour odourless Melting point / freezing point not available Initial boiling point not available Flammability not flammable Lower explosive limit not available Upper explosive limit not available Flash point not available Auto-ignition temperature not available Decomposition temperature not available

Kinematic viscosity not available

Solubility Completely missable with

water

Partition coefficient: n-octanol/water not available Vapour pressure not available

Density and/or relative density

Relative vapour density not available Reason for missing data:Not available

Particle characteristics not applicable

9.2. Other information

9.2.1. Information with regard to physical hazard classes

Information not available

9.2.2. Other safety characteristics

Explosive properties not applicable Oxidising properties not applicable

@EPY 11.4.0 - SDS 1004.14

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SECTION 9. Physical and chemical properties .../>>

Solubility in solvents

solubile in molti solventi organici

SECTION 10. Stability and reactivity

In the absence of information on the mixture, literature information is reported on the components. This information is not characteristic of the solution but of dangerous components.

10.1. Reactivity

Corrosive in contact with metals.

HYDROCHLORIC ACID

The water solution is a strong acid (IPCS, 2000).

By decomposition it develops hydrogen.

Mercury chloride ico

Sodium (na). Potassium (K). In the presence of organic compounds and sunlight, it is slowly decomposed in metal hg. The thermal decomposition generates:

corrosive vapors.

10.2. Chemical stability

There are no particular reaction dangers with other substances in normal conditions of use.

HYDROCHI ORIC ACID

The aqueous solutions are stable

10.3. Possibility of hazardous reactions

HYDROCHLORIC ACID

It reacts violently with bases and obsidants, developing toxic gaseous chlorine (IPCS, 2000). Attack many metals in the presence of water. This produces flammable/explosive gaseous hydrogen (IPCS, 2000).

Mercury chloride ico

Finely dispersed particles form explosive mixtures with air.

Combustion produces harmful and toxic fumes.

10.4. Conditions to avoid

HYDROCHLORIC ACID

High temperature

Mercury chloride ico

Keep away from heat sources and flan sources.

Avoid the formation of dust.

Avoid humidity.

10.5. Incompatible materials

HYDROCHLORIC ACID

Strong oxidant agents, bases, fluorine, amines, reducing agents, metals.

Mercury chloride ico

Strong acids. Strong alkalis. Organic compounds. Formiati, sulfuri, phosphates, albumin, ammonia, gelatin, carbonates, hypofosphites, sulphurous, alkali, alkaloid salts, lime water, antimony and arsenic bromide, bromuries, borace, reduced iron, copper, iron, lead, tannic acid and vegetable astringents.

10.6. Hazardous decomposition products

HYDROCHLORIC ACID

Hydrochloric acid, chlorine dioxide.

For thermal decomposition or in the event of a fire you can free gases and vapors potentially harmful to health (gaseous hydrochloric acid)

Mercury chloride ico

Fumes. Carbon oxide. Carbon dioxide. The thermal decomposition generates: corrosive vapors. The vapors are denser than the air, extremely

dangerous, mercury vapors.

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SECTION 11. Toxicological information

In the absence of experimental toxicological data on the product itself, any dangers of the product for health have been evaluated on the basis of the properties of the substances contained, according to the criteria provided for by the reference legislation for the classification. Therefore consider the concentration of the individual dangerous substances possibly mentioned in Section 3, to evaluate the toxicological effects deriving from exposure to the product.

11.1. Information on hazard classes as defined in Regulation (EC) No 1272/2008

Metabolism, toxicokinetics, mechanism of action and other information

HYDROCHI ORIC ACID

Following inhalation or ingestion it is quickly separated in H+ and cl- which after entering the circle is eliminated with the urine. The activity of hydrogen chloride is associated with its high solubility in water, where it dissociates almost completely. The hydrogen ion forms with the water ion hydronio, this becomes a donor of a proton that has catalytic properties and is therefore capable of reacting with organic molecules. This explains the ability of hydrogen chloride to induce cellular lesions and necrosis.

Information on likely routes of exposure

HYDROCHLORIC ACID

In the professional field, the main exposure routes are the inhalation and the skin.

The general population can be exposed to inhalation, ingestion, skin and eye contact.

Exposure to aerosols locally involves chemical burns whose gravity is according to the concentration of the solution, the importance of contamination and the duration of contact.

At the skin level, warm and painful erythema can be observed, flops or necrosis. Evolution can be complicated with overinfections, aesthetic or functional sequences.

At the eye level there is immediate pain, tearing, conjunctival hyperemia and often blepharospasm. Sequel can be: conjunctival adhesions, corneal opacity, cataracts, glaucoma and even blindness.

The exposure by inhalation to its vapors or aerosols immediately causes irritation of the respiratory system.

In the form of aerosol, the lesions depend on the sizes of the Airosol particles. You can have ricorrhea, sneezing, sensation of nasal and pharyngeal burning, cough, dyspnea, chest pain. Important complications are laryngeal or bronchospasm edema.

The ingestion of concentrated solutions determines buccal, retrostern and epigastric pain associated with hypersciaralorrhea and frequently bloody vomiting. There is metabolic acidosis and an increase in tissue enzymes due to necrosis, hyperleucocytosis, emolis and hypercloremia.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

HYDROCHLORIC ACID

A quick evaporation of the liquid can cause freezing (IPCS, 2000).

The inhalation of high gas concentrations can cause pneumonia and pulmonary edema with consequent reactive airway syndrome (Rads) (bronchial hyperreactivity). The effects can be delayed (IPCS, 2000).

Exposure to aerosols of solutions leads locally chemical burns whose gravity is according to the concentration of the solution, the importance of contamination and the duration of contact.

At the skin level, warm and painful erythema can be observed, flops or necrosis. Evolution can be complicated with overinfections, aesthetic or functional sequences.

At the eye level there is immediate pain, tearing, conjunctival hyperemia and often blepharospasm. Sequel can be: conjunctival adhesions, corneal opacity, cataracts, glaucoma and even blindness.

The exposure by inhalation to its vapors or aerosols immediately causes irritation of the respiratory system.

In the form of aerosol, the lesions depend on the sizes of the Airosol particles. You can have ricorrhea, sneezing, sensation of nasal and pharyngeal burning, cough, dyspnea, chest pain. Important complications are laryngeal or bronchospasm edema.

Upon cessation of exposure, the symptoms almost always regress, but in some cases you can have delayed pulmonary edema within 48 hours.

Secondary infections are a frequent complication.

At the expense of the respiratory system, in the event of extensive injuries, bronchial hypersecreation and the desquamation of the bronchial mucosa determine truncular obstruction and atheighting. Sequence for the respiratory system are: asthma (books of books), bronchial stenosis, bronchiectasias and pulmonary fibrosis.

The ingestion of concentrated solutions determines buccal, retrostern and epigastric pain associated with hypersciaralorrhea and frequently bloody vomiting. There is metabolic acidosis and an increase in tissue enzymes due to necrosis, hyperleucocytosis, emolis and hypercloremia.

The short-term complications are: esophageal or gastric perforation or digestive bleeding, fistulas, breathing difficulties for laryngeal edema, esophagus-tracheal fistula, shock, disseminated intravascular coagulation.

Long -term complications are: digestive stenosis, in particular esopharyngeal.

The exposure repeated to its vapors or aerosols of aqueous solutions can cause irritative effects: dermatitis and conjunctivitis; ulcerations of the nasal mucosa, buccal, epistaxis and gingivorragias; Dental erosions, chronic bronchitis (inrs, 2010).

Interactive effects

Information not available

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ACUTE TOXICITY

ATE (Inhalation) of the mixture:

ATE (Oral) of the mixture:

Not classified (no significant component)

Not classified (no significant component)

ATE (Dermal) of the mixture:

Not classified (no significant component)

Mercury chloride ico

LD50 (Dermal):

41 mg/kg coniglio
LD50 (Oral):

37 mg/kg topo
LC50 (Inhalation mists/powders):

1 mg/l 96h coniglio

HYDROCHLORIC ACID

RATTO DL50 (oral): 700 mg/kg (inrs, 2010) Rabbit DL50 (skin):> 5010 mg/kg (on, 2010)

Ratto CL50-30 minutes (inhalation): 5.7 - 8.3 mg/l (aerosol) (in, 2010).

Mercury chloride ico LD50 rat: 210 mg/kg.

Toxic for ingestion based on the data available on the substance.

SKIN CORROSION / IRRITATION

Corrosive for the skin

Classification based on the experimental value of the pH

HYDROCHLORIC ACID

Exposure to aerosols locally involves chemical burns whose gravity is according to the concentration of the solution, the importance of contamination and the duration of contact (in, 2010).

At the skin level, warm and painful erythema can be observed, flops or necrosis. Evolution can be complicated with superinfections, aesthetic or functional sequences (inrs. 2010).

In animals concentrations between 3.3% and 17% are irritating for the skin; higher concentrations become corrosive (inrs, 2010).

Mercury chloride ico

Corrosive for the skin based on the available data

SERIOUS EYE DAMAGE / IRRITATION

Causes serious eye damage

HYDROCHLORIC ACID

Exposure to aerosols locally involves chemical burns whose gravity is according to the concentration of the solution, the importance of contamination and the duration of contact (in 2010).

At the eye level there is immediate pain, tearing, conjunctival hyperemia and often blepharospasm. The sequels can be: conjunctival adhesions, corneal opacity, cataracts, glaucoma and even blindness (on, 2010).

In animals, concentrations above 3.3% cause serious eye irritation; Symptoms can include redness, swelling, pain and tears. A prolonged exposure or at higher concentrations induces opacity of the cornea, ulceration and decrease in view with the risk of permanent alteration. The severity of irritation is linked to the duration of the treatment (tears have buffer effect and dilute). In the rabbit, 0.1 ml of an aqueous solution at 10% causes permanent alteration of the vision; The non -irritating concentration is 0.33% (inrs, 2010).

Mercury chloride ico

Corrosive eyepiece on the basis of the available data

RESPIRATORY OR SKIN SENSITISATION

Does not meet the classification criteria for this hazard class

Skin sensitization

HYDROCHLORIC ACID

Negative results in the maximization test in the guinea pig (induction and unleashing: solution to 1 %) and in the test of the swelling of the ear in the mouse (induction at 1 %, 5 %unleashing) (inrs, 2010).

GERM CELL MUTAGENICITY

Does not meet the classification criteria for this hazard class

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HYDROCHLORIC ACID

In single studies, hydrochloric acid has induced chromium-somic mutual changes and aberrations in mammal cells. He also induced chromosomal aberrations in insects and plants. He did not induce bacteria mutations (larc, 1992).

Mercury chloride ico

Ranked as a mutagenic on the basis of the available data.

CARCINOGENICITY

Does not meet the classification criteria for this hazard class

HYDROCHLORIC ACID

In a study on deduction workers of the steel, an excess risk of pulmonary cancer in workers exposed mainly to hydrochloric acid was observed. In the same cohort, a increased risk of lady cancer was observed, however, no analysis was conducted on workers exposed to hydrochloric acid. Three studies control in industrial plants do not indicate any association between exposure to hydrochloric acid and cancer against lungs, brain or kidneys. A Canadian control case indicates increased risk for microcytoma in workers exposed to hydrochloric acid; However, no excess risk was observed for other histological types of lung cancer (larc, 1992). In a study in rats m. Exposed for inhalation for the entire life of life at a dose level, hydrogen chloride has not been observed related increase in treatment in the incidence of tumors (larc, 1992).

The International Agency for Research on Cancer (IARC) allocates hydrochloric acid in group 3 (not classifiable as a carcinogen for humans) on the basis of evidence of inadequate carcinogenicity both in man and animals (larc, 1992).

In addition, in a recent evaluation, the data showed association between exposure to mists of strong inorganic acids and laryngeal cancer in the man while they were limited to affirm a causal association with bronchial cancer. The man was also observed in the positive association between exposure to mists of strong inorganic acids and lung cancer (larc, 2012)

- The International Agency for Research on Cancer (IARC) allocates the mists of strong inorganic acids in group 1 (carcinogen ascertained for humans) on the basis of evidence of sufficient carcinogenicity in humans (cancer against the larynx and positive association between exposure to mists of strong inorganic acids and lung cancer) (larc, 2012).

REPRODUCTIVE TOXICITY

Does not meet the classification criteria for this hazard class

Mercury chloride ico

Classified as a reprotoxic based on the available data.

Adverse effects on sexual function and fertility

HYDROCHLORIC ACID

There are no data on humans that allow to evaluate the effects for reproduction of hydrogen chloride exposure. These effects do not seem plausible in professional exposure conditions (in, 2010).

In rats (females) exposed to hydrochloric acid at 450 mg/m3 for 1 hour, both 12 days before the mating and on the 9th day of gestation, effects were observed only at toxic concentrations for mothers (in the, 2010).

Reliable studies on toxicity for reproduction and development in animals are not available after oral, skin or inhalation exposure to hydrochloric acid. Since protons and chloride ions are normal constituents in the body fluids of animal species, low concentrations of gas/mists or of hydrochloric acid solutions do not seem to cause adverse effects in animals. In fact, the cells of gastric glands secrete hydrochloric acid in the cavity of the stomach and also the oral administration of sulfuric acid which determines alteration of the pH, have not caused toxicity on the development in laboratory animals. These facts indicate that the hydrogen chloride and hydrogen chloride present toxicity for development is not expected. Furthermore, in a 90 -day inhalation study of good quality, concentrations up to 50 ppm of the substance did not produce any effect on the Gonadi (Oecd, 2002).

Adverse effects on development of the offspring

HYDROCHLORIC ACID

There are no data on humans that allow to evaluate the effects for reproduction of hydrogen chloride exposure. These effects do not seem plausible in professional exposure conditions (in, 2010).

In rats (females) exposed to hydrochloric acid at 450 mg/m3 for 1 hour, both 12 days before the mating and on the 9th day of gestation, effects were observed only at conc. Toxic for mothers (inrs, 2010).

Reliable studies on toxicity for reproduction and development in animals are not available after oral, skin or inhalation exposure to hydrochloric acid. Since protons and chloride ions are normal constituents in the body fluids of animal species, low conc. of gas/mists or hydrochloric acid solutions do not seem to cause adverse effects in animals. In fact, the cells of gastric glands secrete hydrochloric acid in the cavity of the stomach and also the oral administration of sulfuric acid which determines alteration of the pH, have not caused toxicity on the development in laboratory animals.

These facts indicate that the hydrogen chloride and hydrogen chloride present toxicity for development is not expected. Furthermore, in a 90 -day inhalation study of good quality, conc. Up to 50 ppm of the substance did not produce any effect on the Gonadi (Oecd, 2002).

STOT - SINGLE EXPOSURE

Does not meet the classification criteria for this hazard class

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SECTION 11. Toxicological information .../>>

HYDROCHLORIC ACID

The exposure by inhalation immediately causes irritation of the respiratory system (inrs, 2010).

In the form of aerosol, the lesions depend on the sizes of the Airosol particles. You can have ricorrhea, sneezing, sensation of nasal and pharyngeal burning, cough, dyspnea, chest pain. Important complications are laryngeal or bronchospasm edema (inrs, 2010).

Route of exposure

Mercury chloride ico

Exposure route: inhalation (dust/mists/fumes)

Evaluation: it can irritate the respiratory tract.

STOT - REPEATED EXPOSURE

Does not meet the classification criteria for this hazard class

HYDROCHLORIC ACID

The exposure by inhalation immediately causes irritation of the respiratory system (inrs, 2010).

In the form of aerosol, the lesions depend on the sizes of the Airosol particles. You can have ricorrhea, sneezing, sensation of nasal and pharyngeal burning, cough, dyspnea, chest pain. Important complications are laryngeal or bronchospasm edema (inrs, 2010).

Mercury chloride ico

Ranked for this end-point based on the available data

ASPIRATION HAZARD

Does not meet the classification criteria for this hazard class

11.2. Information on other hazards

Based on the available data, the product does not contain substances listed in the main European lists of potential or suspected endocrine disruptors with human health effects under evaluation.

SECTION 12. Ecological information

Use according to good working practices, avoiding dispersing the product in the environment. Notify the competent authorities if the product has reached waterways or if it has contaminated the soil or vegetation.

12.1. Toxicity

HYDROCHLORIC ACID

Short -term effects

Fish (Cyprinus Carpio) CL50-96 hours: 4.92 mg/l a pH 4.3 [OECD 203] (Oecd Sids, 2002).

Fish (Gambusia Affinis) CL50-96 hours = 282 mg/l (pH 6.0-8,2) (HSDB, 2015).

Crustaceans (Daphnia Magna) Ce50-48 hours = 0.492 mg/l (Ph 5,3) [Oecd 202] (Oecd Sids, 2002).

Alga (pseudokirkinchneriella subcapitated) cbe50-72 hours = 0.780 mg/l (pH 5.1); Cbe50-72 hours = 0.492 mg/l (Ph 5,3) [Oecd 201] (Oecd Sids, 2002).

In the air it can be phytotoxic.

Tomatoes, sugar beets and some fruit trees are sensitive to hydrogen chloride in the air (HSDB, 2015).

The aqueous solutions of hydrochloric acid have a corrosive action on plant tissues.

Long -term effects

Algae (pseudokirkinchneriella subcapitated) Noec = 0.097 mg/l (Ph 6.0) [Oecd 201] (effect: growth rate and biomass) (Oecd Sids, 2002).

Mercury chloride ico

Pisces toxicity - Loec mortality - Lates Calcarifer - 0.113 mg/l - 96 h

Algae toxicity - CE50 growth inhibitor - Ditylum Brightwellii - 0.01 mg/l - 5 days

Mercury chloride ico

LC50 - for Fish
EC50 - for Crustacea

0,016 mg/l/96h Oncorhynchus mykiss (Trota iridea) 0,002 mg/l/48h Daphnia magna (Pulce d'acqua grande)

12.2. Persistence and degradability

HYDROCHLORIC ACID

In the water it dissociates.

The substance is not photo -degradable.

12.3. Bioaccumulative potential

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SECTION 12. Ecological information .../>>

HYDROCHLORIC ACID

Biochoncentration is not significant.

BCF given not available.

Mercury chloride ico

Pimephales Promolas (American Cave) 0.50 Microg/L

Mercury chloride ico

BCF 5,68

12.4. Mobility in soil

HYDROCHLORIC ACID It is mobile on the ground

12.5. Results of PBT and vPvB assessment

On the basis of available data, the product does not contain any PBT or vPvB in percentage ≥ than 0,1%.

12.6. Endocrine disrupting properties

HYDROCHLORIC ACID

Despite the dilution, it shapes with water still corrosive mixtures. Harmful effect due to the variation of the pH.

Based on the available data, the product does not contain substances listed in the main European lists of potential or suspected endocrine disruptors with environmental effects under evaluation.

12.7. Other adverse effects

Information not available

SECTION 13. Disposal considerations

13.1. Waste treatment methods

Reuse, when possible. Product residues should be considered special hazardous waste. The hazard level of waste containing this product should be evaluated according to applicable regulations.

Disposal must be performed through an authorised waste management firm, in compliance with national and local regulations.

Waste transportation may be subject to ADR restrictions.

CONTAMINATED PACKAGING

Contaminated packaging must be recovered or disposed of in compliance with national waste management regulations.

SECTION 14. Transport information

14.1. UN number or ID number

ADR / RID, IMDG, IATA: 1789

14.2. UN proper shipping name

ADR / RID: HYDROCHLORIC ACID IMDG: HYDROCHLORIC ACID IATA: HYDROCHLORIC ACID

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SECTION 14. Transport information .../>>

14.3. Transport hazard class(es)

ADR / RID: Class: 8 Label: 8

IMDG: Class: 8 Label: 8

IATA: Class: 8 Label: 8



14.4. Packing group

ADR / RID, IMDG, IATA: III

14.5. Environmental hazards

ADR / RID: NO IMDG: NO IATA: NO

14.6. Special precautions for user

ADR / RID: HIN - Kemler: 80 Limited Quantities: 5 L Tunnel restriction code: (E)

Special provision: 520 IMDG: EMS: F-A, S-B

 IMDG:
 EMS: F-A, S-B
 Limited Quantities: 5 L

 IATA:
 Cargo:
 Maximum quantity: 60 L
 Packaging instructions: 856

 Pass.:
 Maximum quantity: 5 L
 Packaging instructions: 852

Special provision: A3, A803

14.7. Maritime transport in bulk according to IMO instruments

Information not relevant

SECTION 15. Regulatory information

15.1. Safety, health and environmental regulations/legislation specific for the substance or mixture

Seveso Category - Directive 2012/18/EU: None

Restrictions relating to the product or contained substances pursuant to Annex XVII to EC Regulation 1907/2006

<u>Product</u>

Point 3
Contained substance

Point 75
Point 18 Mercury chloride ico

Regulation (EU) 2019/1148 - on the marketing and use of explosives precursors

not applicable

Substances in Candidate List (Art. 59 REACH)

On the basis of available data, the product does not contain any SVHC in percentage ≥ than 0,1%.

Substances subject to authorisation (Annex XIV REACH)

None

Substances subject to exportation reporting pursuant to Regulation (EU) 649/2012:

None

Substances subject to the Rotterdam Convention:

None

Substances subject to the Stockholm Convention:

None

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SECTION 15. Regulatory information .../>>

Healthcare controls

Workers exposed to this chemical agent must not undergo health checks, provided that available risk-assessment data prove that the risks related to the workers' health and safety are modest and that the 98/24/EC directive is respected.

15.2. Chemical safety assessment

A chemical safety assessment has been performed for the following contained substances HYDROCHLORIC ACID

SECTION 16. Other information

Text of hazard (H) indications mentioned in section 2-3 of the sheet:

Met. Corr. 1 Substance or mixture corrosive to metals, category 1

Muta. 2 Germ cell mutagenicity, category 2
Repr. 2 Reproductive toxicity, category 2
Acute Tox. 2 Acute toxicity, category 2

STOT RE 1 Specific target organ toxicity - repeated exposure, category 1

Skin Corr. 1B Skin corrosion, category 1B
Eye Dam. 1 Serious eye damage, category 1

STOT SE 3 Specific target organ toxicity - single exposure, category 3

Aquatic Acute 1 Hazardous to the aquatic environment, acute toxicity, category 1

Aquatic Chronic 1 Hazardous to the aquatic environment, chronic toxicity, category 1

H290 May be corrosive to metals.

H341 Suspected of causing genetic defects.

H361 Suspected of damaging fertility or the unborn child.

H300 Fatal if swallowed.

H372 Causes damage to organs through prolonged or repeated exposure.

H314 Causes severe skin burns and eye damage.

H318 Causes serious eye damage.
 H335 May cause respiratory irritation.
 H400 Very toxic to aquatic life.

H410 Very toxic to aquatic life with long lasting effects.

LEGEND:

- ADR: European Agreement concerning the carriage of Dangerous goods by Road
- ATE: Acute Toxicity Estimate
- CAS: Chemical Abstract Service Number
- CE50: Effective concentration (required to induce a 50% effect)
- CE: Identifier in ESIS (European archive of existing substances)
- CLP: Regulation (EC) 1272/2008
- DNEL: Derived No Effect Level
- EmS: Emergency Schedule
- GHS: Globally Harmonized System of classification and labeling of chemicals
- IATA DGR: International Air Transport Association Dangerous Goods Regulation
- IC50: Immobilization Concentration 50%
- IMDG: International Maritime Code for dangerous goods
- IMO: International Maritime Organization
- INDEX: Identifier in Annex VI of CLP
- LC50: Lethal Concentration 50%
- LD50: Lethal dose 50%
- OEL: Occupational Exposure Level
- PBT: Persistent bioaccumulative and toxic as REACH Regulation
- PEC: Predicted environmental Concentration
- PEL: Predicted exposure level
- PNEC: Predicted no effect concentration
- REACH: Regulation (EC) 1907/2006
- RID: Regulation concerning the international transport of dangerous goods by train
- TLV: Threshold Limit Value
- TLV CEILING: Concentration that should not be exceeded during any time of occupational exposure.
- TWA: Time-weighted average exposure limit
- TWA STEL: Short-term exposure limit
- VOC: Volatile organic Compounds
- vPvB: Very Persistent and very Bioaccumulative as for REACH Regulation
- WGK: Water hazard classes (German).

GENERAL BIBLIOGRAPHY

- 1. Regulation (EC) 1907/2006 (REACH) of the European Parliament
- 2. Regulation (EC) 1272/2008 (CLP) of the European Parliament
- 3. Regulation (EU) 2020/878 (II Annex of REACH Regulation)

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- 4. Regulation (EC) 790/2009 (I Atp. CLP) of the European Parliament
- 5. Regulation (EU) 286/2011 (II Atp. CLP) of the European Parliament
- 6. Regulation (EU) 618/2012 (III Atp. CLP) of the European Parliament
- 7. Regulation (EU) 487/2013 (IV Atp. CLP) of the European Parliament
- 8. Regulation (EU) 944/2013 (V Atp. CLP) of the European Parliament
- 9. Regulation (EU) 605/2014 (VI Atp. CLP) of the European Parliament
- 10. Regulation (EU) 2015/1221 (VII Atp. CLP) of the European Parliament
- 11. Regulation (EU) 2016/918 (VIII Atp. CLP) of the European Parliament
- 12. Regulation (EU) 2016/1179 (IX Atp. CLP)
- 13. Regulation (EU) 2017/776 (X Atp. CLP)
- 14. Regulation (EU) 2018/669 (XI Atp. CLP)
- 15. Regulation (EU) 2019/521 (XII Atp. CLP)
- 16. Delegated Regulation (UE) 2018/1480 (XIII Atp. CLP)
- 17. Regulation (EU) 2019/1148
- 18. Delegated Regulation (UE) 2020/217 (XIV Atp. CLP)
- 19. Delegated Regulation (UE) 2020/1182 (XV Atp. CLP)
- 20. Delegated Regulation (UE) 2021/643 (XVI Atp. CLP)
- 21. Delegated Regulation (UE) 2021/849 (XVII Atp. CLP)
- 22. Delegated Regulation (UE) 2022/692 (XVIII Atp. CLP)
- The Merck Index. 10th Edition
- Handling Chemical Safety
- INRS Fiche Toxicologique (toxicological sheet)
- Patty Industrial Hygiene and Toxicology
- N.I. Sax Dangerous properties of Industrial Materials-7, 1989 Edition
- IFA GESTIS website
- ECHA website
- Database of SDS models for chemicals Ministry of Health and ISS (Istituto Superiore di Sanità) Italy

Note for users:

The information contained in the present sheet are based on our own knowledge on the date of the last version. Users must verify the suitability and thoroughness of provided information according to each specific use of the product.

This document must not be regarded as a guarantee on any specific product property.

The use of this product is not subject to our direct control; therefore, users must, under their own responsibility, comply with the current health and safety laws and regulations. The producer is relieved from any liability arising from improper uses.

Provide appointed staff with adequate training on how to use chemical products.

CALCULATION METHODS FOR CLASSIFICATION

Chemical and physical hazards: Product classification derives from criteria established by the CLP Regulation, Annex I, Part 2. The data for evaluation of chemical-physical properties are reported in section 9.

Health hazards: Product classification is based on calculation methods as per Annex I of CLP, Part 3, unless determined otherwise in Section 11

Environmental hazards: Product classification is based on calculation methods as per Annex I of CLP, Part 4, unless determined otherwise in Section 12.

Changes to previous review:

The following sections were modified:

03 / 04 / 07 / 10.